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Lakhmir Singh Solutions Class 10 Chemistry Chapter 2 Acids, Bases and Salts On the basis of the chemical property of compounds, it has been divided into three groups

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Acids, bases and salts where acids and bases combined to form salts.

[Acids Bases and Salts Class 10 Notes Science Chapter 2](#)

Mineral Acids - Acids that prepare from mineral are known as mineral acids/ inorganic acids/man-made acids or synthetic acid, such as hydrochloric acid (HCl), sulphuric acid (H_2SO_4), nitric acid (HNO_3), etc. Note: Organic acids are always weak but minerals acids can be strong as well as weak. We can take dilute weak acids (like organic acid) in our body.

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Inorganic compounds essential to

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human functioning include water, salts, acids, and bases. These compounds are inorganic; that is, they do not contain both hydrogen and carbon. Water is a lubricant and cushion, a heat sink, a component of liquid mixtures, a byproduct of dehydration synthesis reactions, and a reactant in hydrolysis reactions.

[14.4 Hydrolysis of Salts - Chemistry 2e | OpenStax](#)

Inorganic acids and inorganic oxygen compounds of non-metals; Halogen or sulphur compounds of non-metals; Inorganic bases and oxides, hydroxides and peroxides of metals; Salts and peroxy salts, of inorganic acids and metals; Miscellaneous;

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Chapter 29 - Organic chemicals

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